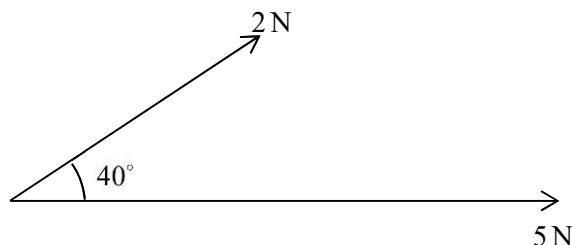


Section B: Mechanics
Answer **all** the questions

- 9 Two forces, of magnitudes 2 N and 5 N, act on a particle in the directions shown in the diagram below.



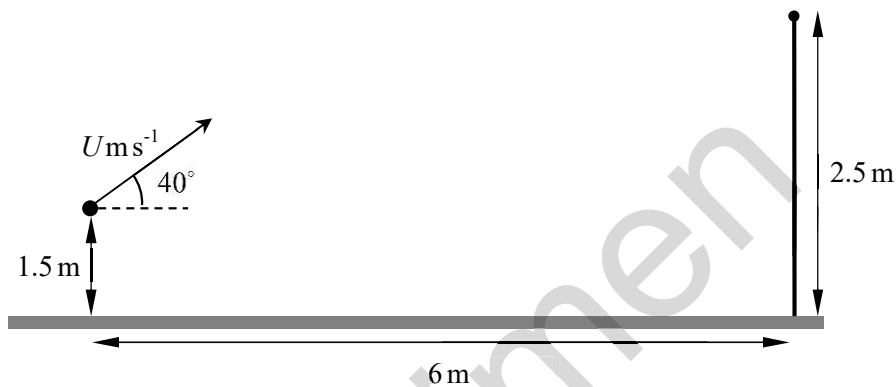
- (i) Calculate the magnitude of the resultant force on the particle. [3]
- (ii) Calculate the angle between this resultant force and the force of magnitude 5 N. [1]
- 10 A body of mass 20 kg is on a rough plane inclined at angle α to the horizontal. The body is held at rest on the plane by the action of a force of magnitude P N acting up the plane in a direction parallel to a line of greatest slope of the plane. The coefficient of friction between the body and the plane is μ .
- (i) When $P = 100$, the body is on the point of sliding down the plane. Show that $g \sin \alpha = g \mu \cos \alpha + 5$. [4]
- (ii) When P is increased to 150, the body is on the point of sliding up the plane. Using this and your answer to part (i), find an expression for α in terms of g . [3]
- 11 In this question the unit vectors \mathbf{i} and \mathbf{j} are in the directions east and north respectively.
- A particle of mass 0.12 kg is moving so that its position vector \mathbf{r} metres at time t seconds is given by $\mathbf{r} = 2t^3\mathbf{i} + (5t^2 - 4t)\mathbf{j}$.
- (i) Show that when $t = 0.7$ the bearing on which the particle is moving is approximately 044° . [3]
- (ii) Find the magnitude of the resultant force acting on the particle at the instant when $t = 0.7$. [4]
- (iii) Determine the times at which the particle is moving on a bearing of 045° . [2]

- 12 A girl is practising netball. She throws the ball from a height of 1.5 m above horizontal ground and aims to get the ball through a hoop. The hoop is 2.5 m vertically above the ground and is 6 m horizontally from the point of projection.

The situation is modelled as follows.

- The initial velocity of the ball has magnitude $U \text{ m s}^{-1}$.
- The angle of projection is 40° .
- The ball is modelled as a particle.
- The hoop is modelled as a point.

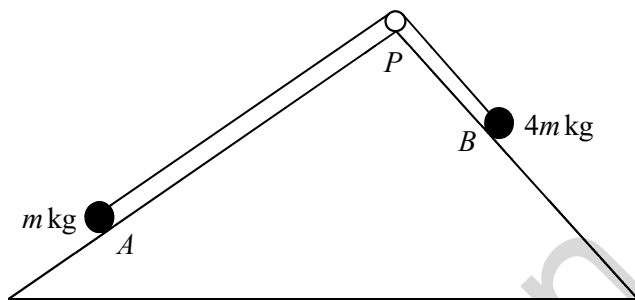
This is shown on the diagram below.



- (i) For $U = 10$, find
- (a) the greatest height above the ground reached by the ball, [5]
- (b) the distance between the ball and the hoop when the ball is vertically above the hoop. [4]
- (ii) Calculate the value of U which allows her to hit the hoop. [3]
- (iii) How appropriate is this model for predicting the path of the ball when it is thrown by the girl? [1]
- (iv) Suggest one improvement that might be made to this model. [1]

- 13 Particle A , of mass m kg, lies on the plane Π_1 inclined at an angle of $\tan^{-1} \frac{3}{4}$ to the horizontal. Particle B , of $4m$ kg, lies on the plane Π_2 inclined at an angle of $\tan^{-1} \frac{4}{3}$ to the horizontal. The particles are attached to the ends of a light inextensible string which passes over a smooth pulley at P . The coefficient of friction between particle A and Π_1 is $\frac{1}{3}$ and plane Π_2 is smooth. Particle A is initially held at rest such that the string is taut and lies in a line of greatest slope of each plane.

This is shown on the diagram below.



- (i) Show that when A is released it accelerates towards the pulley at $\frac{7g}{15} \text{ ms}^{-2}$. [6]
- (ii) Assuming that A does not reach the pulley, show that it has moved a distance of $\frac{1}{4}$ m when its speed is $\sqrt{\frac{7g}{30}} \text{ ms}^{-1}$. [2]
- 14 A uniform ladder AB of mass 35 kg and length 7 m rests with its end A on rough horizontal ground and its end B against a rough vertical wall. The ladder is inclined at an angle of 45° to the horizontal. A man of mass 70 kg is standing on the ladder at a point C , which is x metres from A . The coefficient of friction between the ladder and the wall is $\frac{1}{3}$ and the coefficient of friction between the ladder and the ground is $\frac{1}{2}$. The system is in limiting equilibrium. Find x . [8]

- 1 Express $\frac{8}{\sqrt{3}-1}$ in the form $a\sqrt{3}+b$, where a and b are integers. [3]
- 2 (i) Sketch the curve $y = -\frac{1}{x}$. [2]
- (ii) The curve $y = -\frac{1}{x}$ is translated by 2 units parallel to the x -axis in the positive direction. State the equation of the transformed curve. [2]
- (iii) Describe a transformation that transforms the curve $y = -\frac{1}{x}$ to the curve $y = -\frac{1}{3x}$. [2]
- 3 Express each of the following in the form 5^k .
- (i) 25^4 [1]
- (ii) $\frac{1}{\sqrt[4]{5}}$ [2]
- (iii) $(5\sqrt{5})^3$ [2]
- 4 Solve the equation $x^{\frac{2}{3}} - x^{\frac{1}{3}} - 6 = 0$. [5]
- 5 The points A and B have coordinates $(2, 1)$ and $(5, -3)$ respectively.
- (i) Find the length of AB . [2]
- (ii) Find an equation of the line through the mid-point of AB which is perpendicular to AB , giving your answer in the form $ax + by + c = 0$ where a , b and c are integers. [7]
- 6 Solve the simultaneous equations
- $$2x + y - 5 = 0, \quad x^2 - y^2 = 3. \quad [5]$$
- 7 (a) Given that $f(x) = (x^2 + 3)(5 - x)$, find $f'(x)$. [4]
- (b) Find the gradient of the curve $y = x^{-\frac{1}{3}}$ at the point where $x = -8$. [4]

- 8 (i) Sketch the curve $y = 2x^2 - x - 3$, giving the coordinates of all points of intersection with the axes. [4]
- (ii) Hence, or otherwise, solve the inequality $2x^2 - x - 3 > 0$. [2]
- (iii) Given that the equation $2x^2 - x - 3 = k$ has no real roots, find the set of possible values of the constant k . [3]
- 9 The curve $y = 2x^3 - ax^2 + 8x + 2$ passes through the point B where $x = 4$.
- (i) Given that B is a stationary point of the curve, find the value of the constant a . [5]
- (ii) Determine whether the stationary point B is a maximum point or a minimum point. [2]
- (iii) Find the x -coordinate of the other stationary point of the curve. [3]
- 10 A circle with centre C has equation $x^2 + y^2 - 10x + 4y + 4 = 0$.
- (i) Find the coordinates of C and the radius of the circle. [3]
- (ii) Show that the tangent to the circle at the point $P(8, 2)$ has equation $3x + 4y = 32$. [5]
- (iii) The circle meets the y -axis at Q and the tangent meets the y -axis at R . Find the area of triangle PQR . [4]

1. Balance the following chemical reactions:
 - a. $\text{CO} + \text{O}_2 \rightarrow \text{CO}_2$
 - b. $\text{KNO}_3 \rightarrow \text{KNO}_2 + \text{O}_2$
 - c. $\text{O}_3 \rightarrow \text{O}_2$
 - d. $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \text{H}_2\text{O}$
 - e. $\text{CH}_3\text{NH}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{N}_2$
 - f. $\text{Cr}(\text{OH})_3 + \text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + \text{H}_2\text{O}$
2. Write the balanced chemical equations of each reaction:
 - a. Calcium carbide (CaC_2) reacts with water to form calcium hydroxide ($\text{Ca}(\text{OH})_2$) and acetylene gas (C_2H_2).
 - b. When potassium chlorate (KClO_3) is heated, it decomposes to form KCl and oxygen gas (O_2).
 - c. C_6H_6 combusts in air.
 - d. $\text{C}_5\text{H}_{12}\text{O}$ combusts in air.
3. Given the following reaction: $\text{Na}_2\text{S}_2\text{O}_3 + \text{AgBr} \rightarrow \text{NaBr} + \text{Na}_3[\text{Ag}(\text{S}_2\text{O}_3)_2]$
 - a. How many moles of $\text{Na}_2\text{S}_2\text{O}_3$ are needed to react completely with 42.7 g of AgBr?
 - b. What is the mass of NaBr that will be produced from 42.7 g of AgBr?
4. From the reaction: $\text{B}_2\text{H}_6 + \text{O}_2 \rightarrow \text{HBO}_2 + \text{H}_2\text{O}$
 - a. What mass of O_2 will be needed to burn 36.1 g of B_2H_6 ?
 - b. How many moles of water are produced from 19.2 g of B_2H_6 ?
5. Calculate the mass (in kg) of water produced from the combustion of 1.0 gallon (3.8 L) of gasoline (C_8H_{18}). The density of gasoline is 0.79 g/mL.
6. One mole of aspartame ($\text{C}_{14}\text{H}_{18}\text{N}_2\text{O}_5$) reacts with two moles of water to produce one mole of aspartic acid ($\text{C}_4\text{H}_7\text{NO}_4$), one mole of methanol (CH_3OH) and one mole of phenylalanine.
 - a. What is the molecular formula of phenylalanine?

- b. What mass of phenylalanine is produced from 378 g of aspartame?
7. KO_2 is used in a closed-system breathing apparatus. It removes carbon dioxide and water from exhaled air. The reaction for the removal of water is: $\text{KO}_2 + \text{H}_2\text{O} \rightarrow \text{O}_2 + \text{KOH}$. The KOH produced is used to remove carbon dioxide by the following reaction: $\text{KOH} + \text{CO}_2 \rightarrow \text{KHCO}_3$.
- a. What mass of KO_2 produces 235 g of O_2 ?
- b. What mass of CO_2 can be removed by 123 g of KO_2 ?

Thermal Reactions

8. How many kilojoules are given off when 17.8 mol of $\text{CH}_4(\text{g})$ react?
 $\text{CH}_4(\text{g}) + 2 \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta H = -890.1 \text{ kJ}$
9. How many kilojoules are absorbed when 23.09 mol of $\text{C}_6\text{H}_6(\text{l})$ are formed?
 $6 \text{C}(\text{s}) + 3 \text{H}_2(\text{g}) \rightarrow \text{C}_6\text{H}_6(\text{l}) \quad \Delta H = 49.0 \text{ kJ}$
-

Electrolysis

1 Faraday (F) = 96 500 Coulombs (C) = 1 mole of electrons.

10. A current was passed through an electrolysis circuit of silver nitrate solution and 0.54g of silver was formed.
- $A_r(\text{Ag}) = 108$ and the electrode equation is $\text{Ag}^+(\text{aq}) + \text{e}^- \Rightarrow \text{Ag}(\text{s})$
 - $A_r(\text{Ag}) = 64$ and the electrode equation is $\text{Cu}^{2+}(\text{aq}) + 2\text{e}^- \Rightarrow \text{Cu}(\text{s})$
 - If in the same circuit a copper(II) sulphate and copper electrodes cell was connected, how much copper is deposited at the negative (-) cathode?
11. How long will it take to produce 2 dm^3 of chlorine gas by passing a 6A current through concentrated sodium chloride solution at 25C and 101kPa (1 atmosphere pressure)
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